

[REDACTED]

[REDACTED]

[REDACTED] 嘉 薩 敏 口

[REDACTED]

[REDACTED] 阿 部 善 行

[REDACTED]

[REDACTED]

Constitution of the R. D. of C. W. S.

[REDACTED]

[REDACTED]

[REDACTED]

[REDACTED]

[REDACTED]

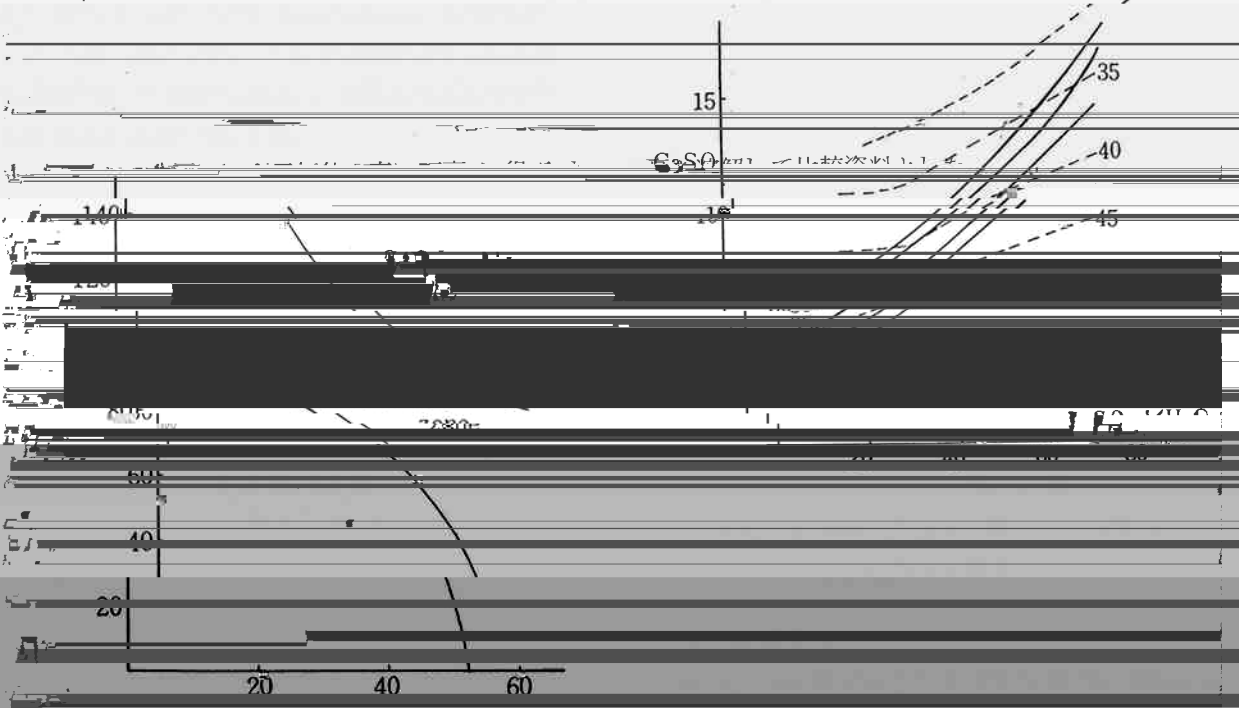
[REDACTED]

[REDACTED]

[REDACTED]

[REDACTED]

[REDACTED]



CaSO₄·2H₂O
Stable zone

$$\therefore A_{\text{avg}} = \frac{A_0^{3/2} - A^{3/2}}{3(A_0^{1/2} - A^{1/2})}$$

$$\Delta_0 - \Delta$$

Table 2

Effect of Excess H₂SO₄

Table 3

2.9 3.0 3.1 3.2

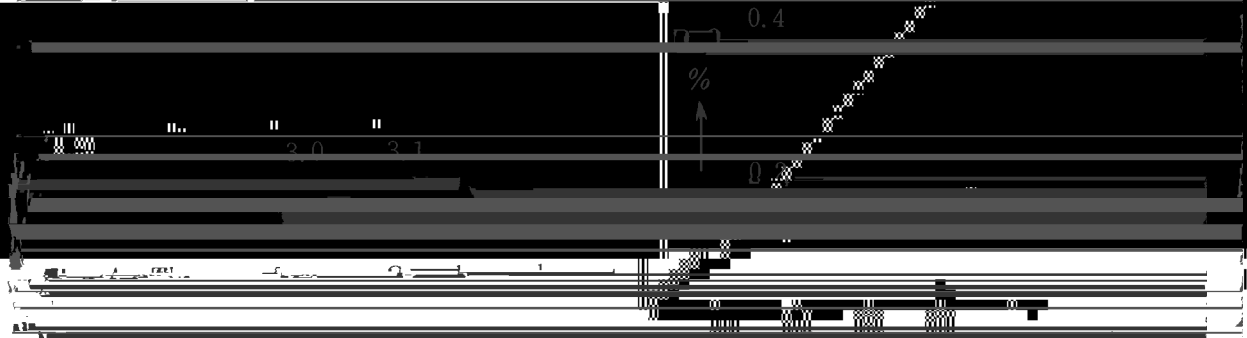
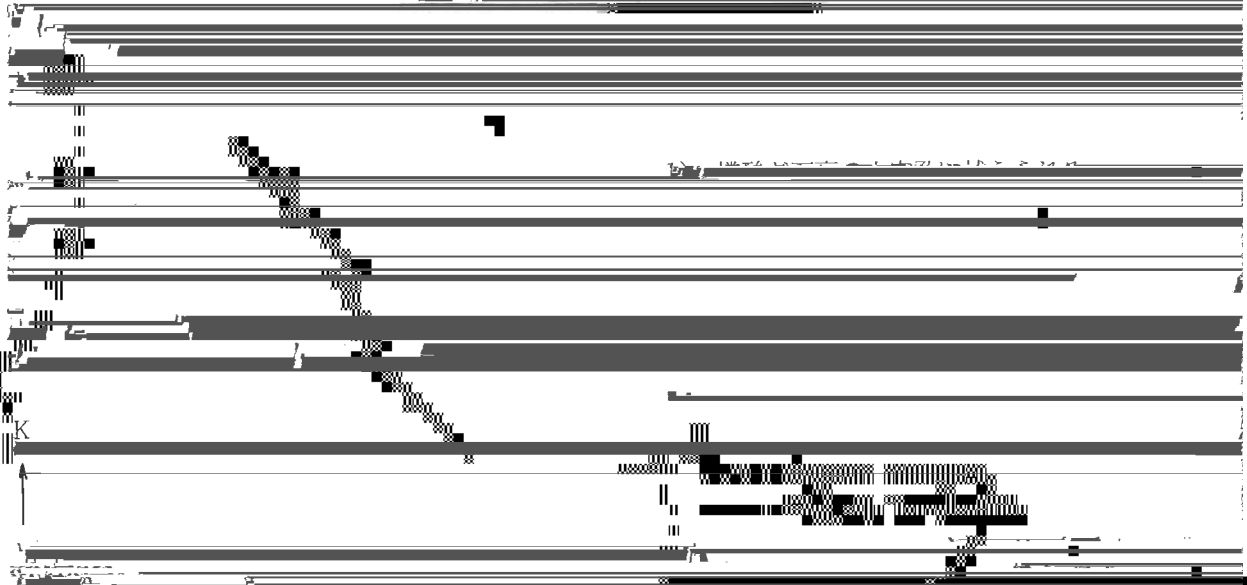
$\times 10^3$

(Re = 4.15×10^4)

3. 成長速度

(本表の成長速度は、その成長速度と比較し、イオン

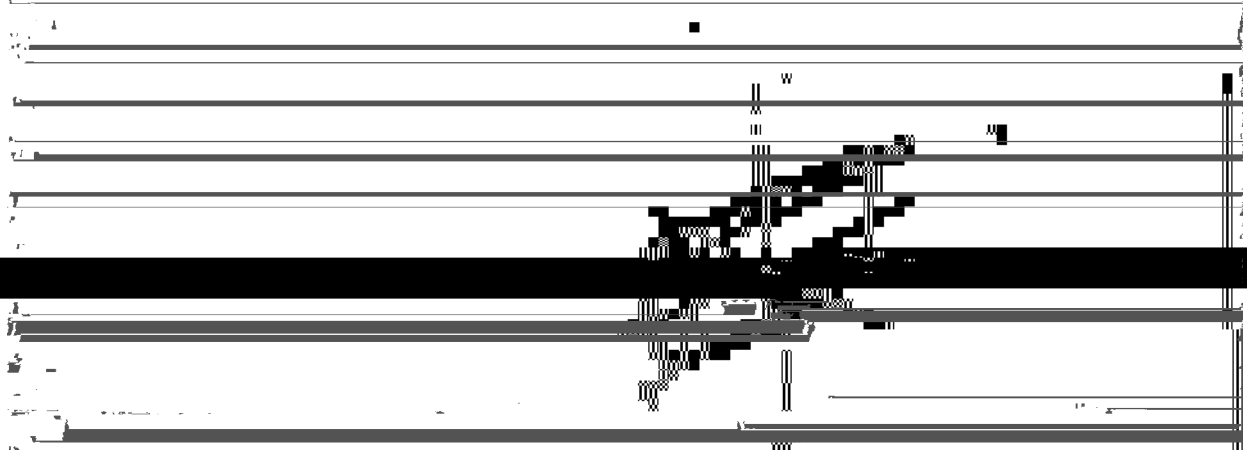
1. 硫酸化処理高の巨山条件



2. 硫酸化処理高の巨山条件

硫酸化処理高の巨山条件

硫酸化処理高の巨山条件



硫酸化処理高の巨山条件

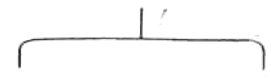
硫酸化処理高の巨山条件

硫酸化処理高の巨山条件

リン酸副生石膏の晶出条件

コーラ
リン鉱石

()内はGypsum (0.308)
硫酸分解



0.016 0.660

(水層) (alcohol層)

Table 5

	0.08		
0.020	0.05		
0	0.007		
	0.035		
	0.009		

0.119 0.000

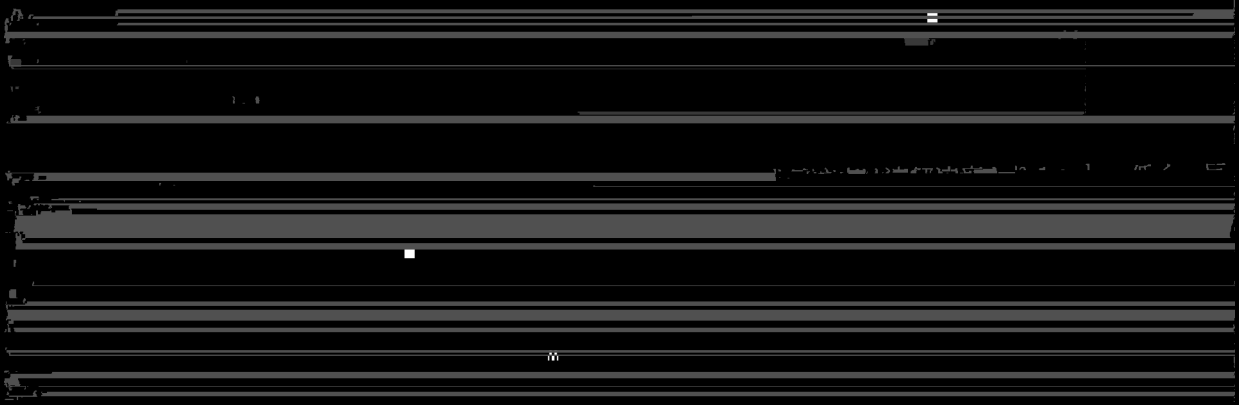
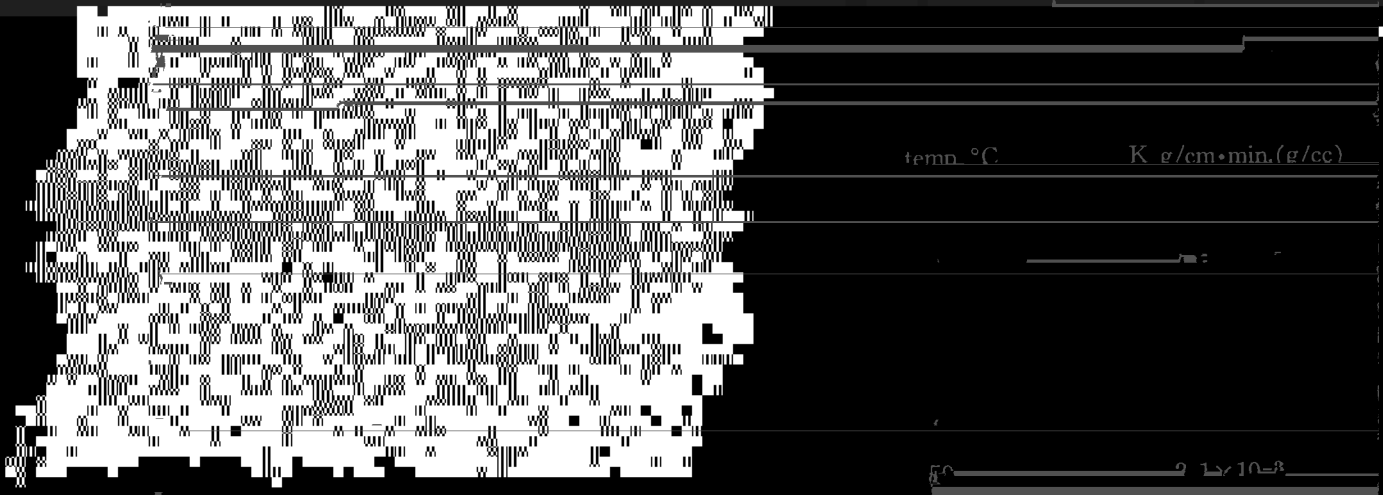
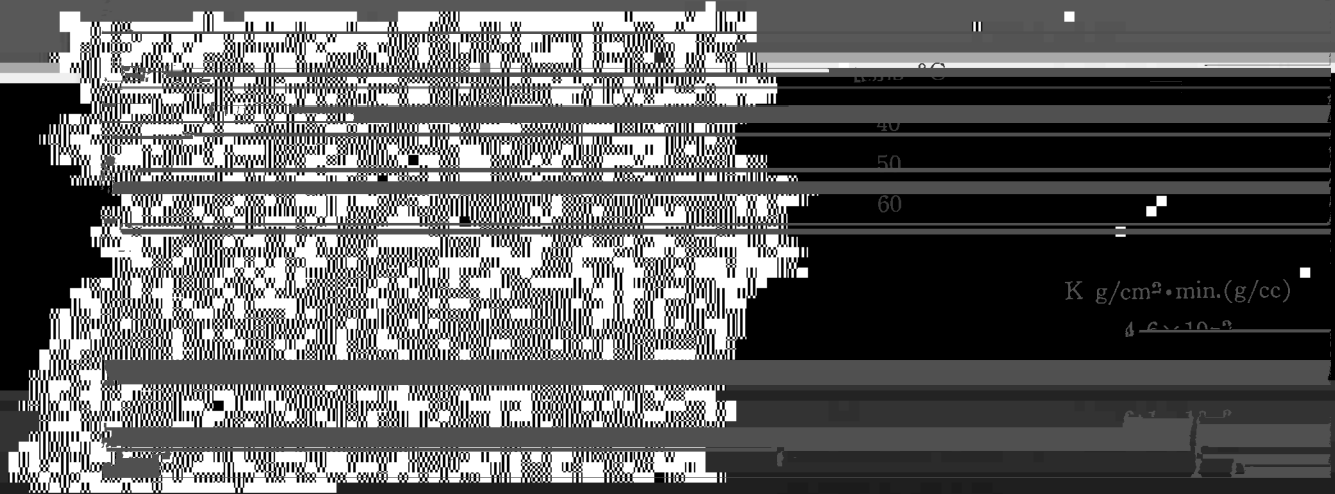
Table 7

		K g/cm ² ·min
60	0	8.4 × 10 ⁻³
60	0.47	6.0 × 10 ⁻³
60	0.68	4.7 × 10 ⁻³

Table 8

Exp. No.	florida-77	99	95	97	97
	Kals-94	1	5	8	3
Crystallization temp °C	60-65	65-70	65-70	65-70	65-70

石膏副産物の品出条件



50